

Submission to Turnitin – Tutor feedback

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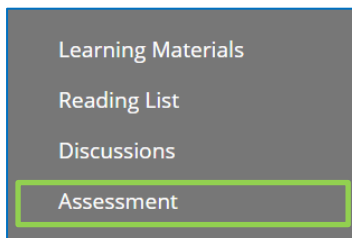
Introduction

To access your tutor feedback, you will need to return to the submission area for that assignment and click on the Turnitin link.



Finding your assignment submission area

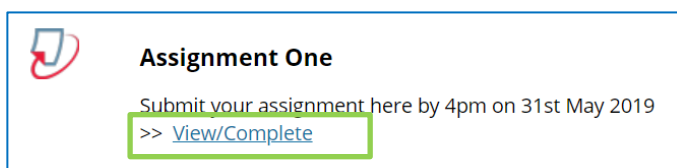
All Blackboard courses should display a link in the course menu that is called either **Assessment** or **Assignments**. This is where you should find all of the assessment information for the module and links to your Turnitin assignment submissions.



Please speak to your tutor(s) if this information or your assignment(s) do not appear here.

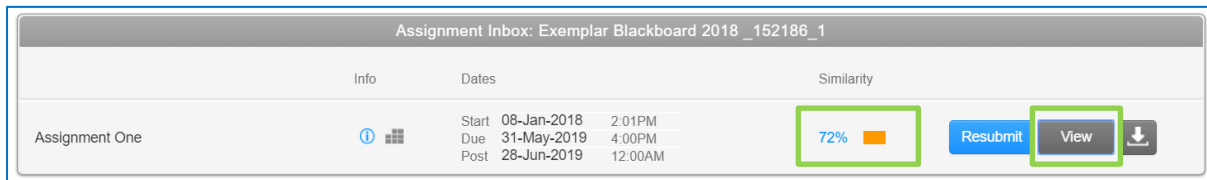
Accessing the Turnitin submission

From the **Assessment** page, click on the **View/Complete** button - as you did when submitting your assignment.



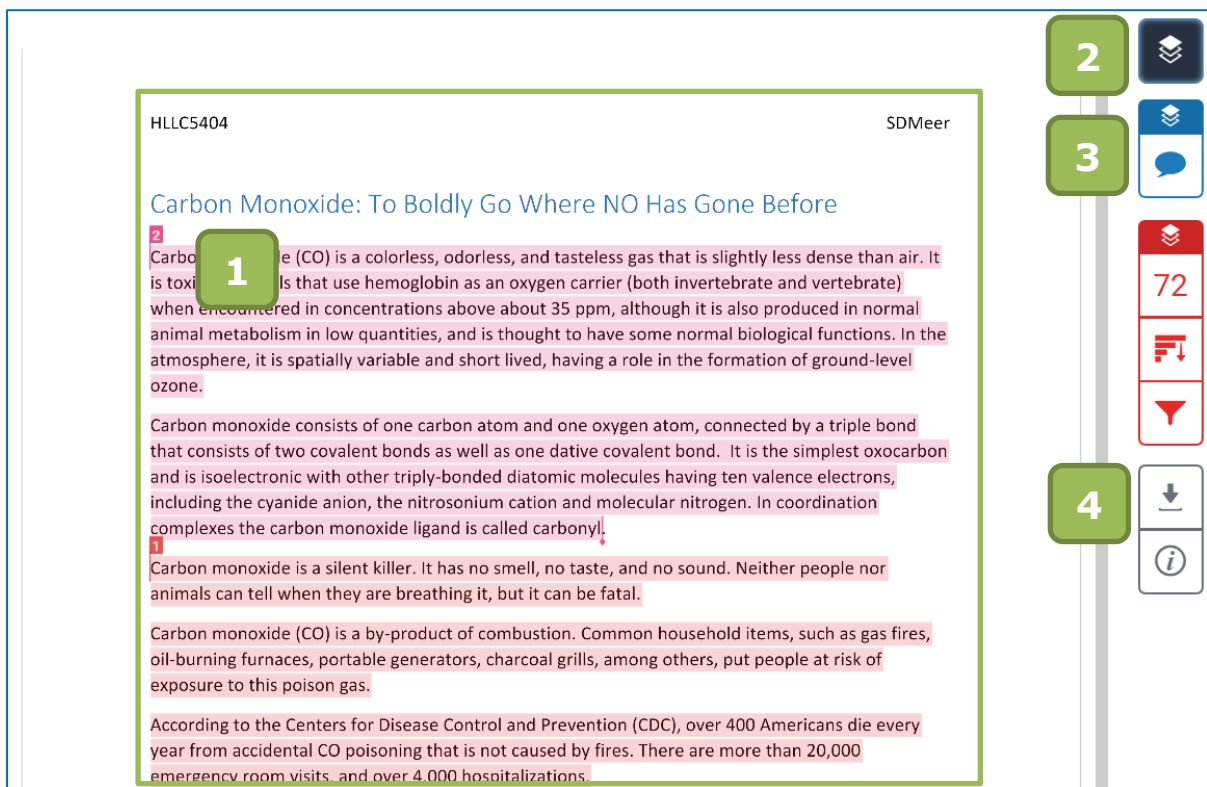
Note: You may see more than one Turnitin submission on the Assessments page – ensure you are selecting the correct one – you may need to scroll down until you see the one you want.

Click on either the **percentage** or the **View** button to see your tutor feedback. Your submitted assignment will open in a new **feedback studio** window.

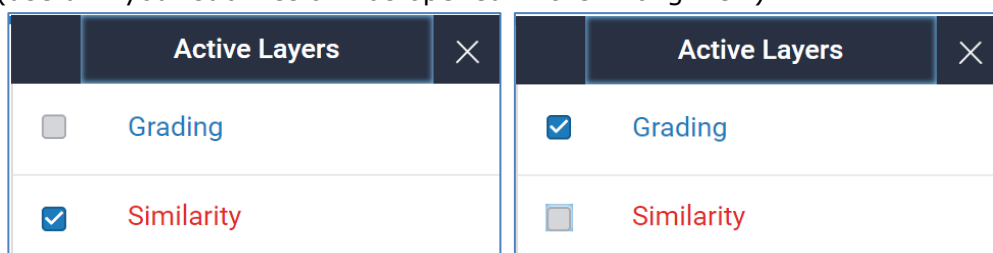


Feedback Studio

With the feedback studio window open, your submitted assignment will usually open in “**similarity view**”. There are a number of things to note in this window:



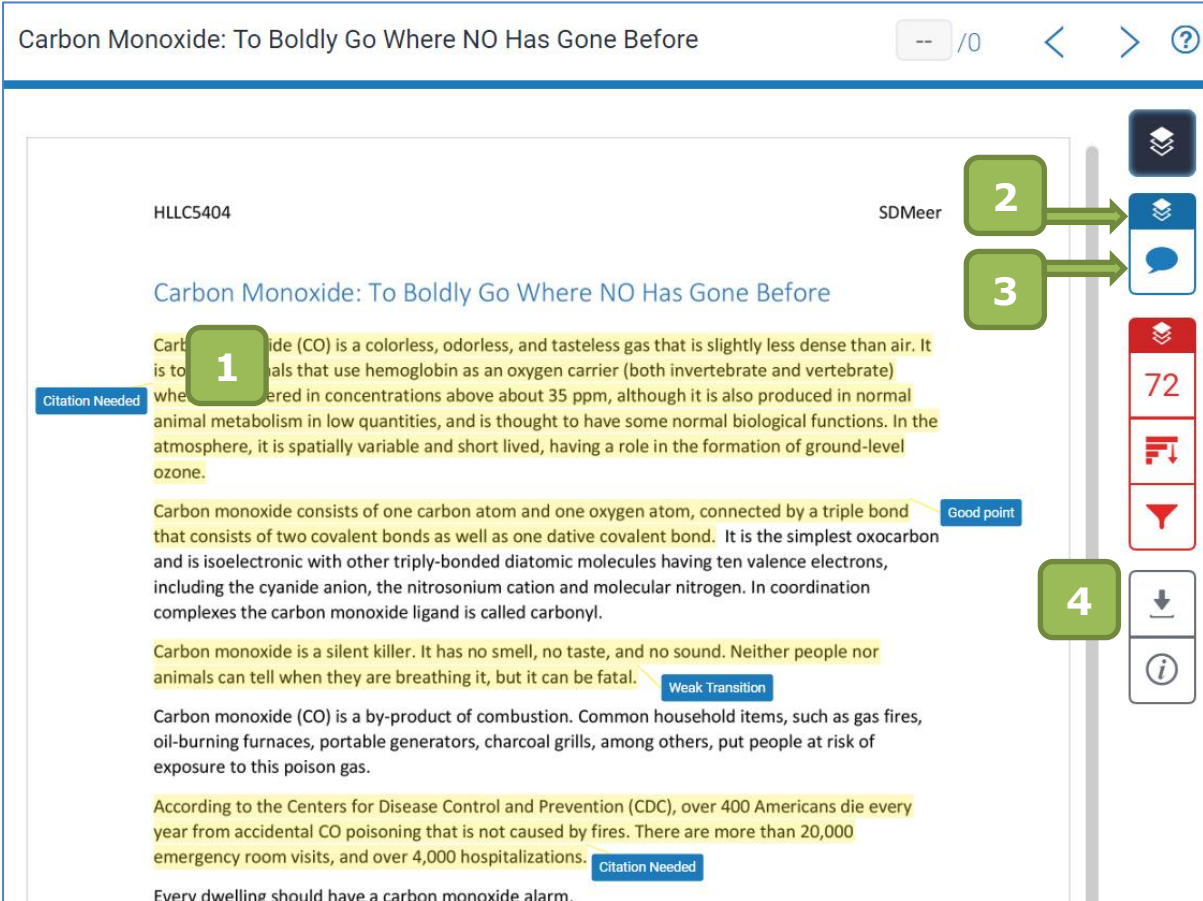
- 1. Your assignment** in similarity view.
- 2. Active Layers** – toggle switches to **turn similarity and grading on and off** (useful if your submission has opened in the wrong view).



- 3. Grading view** is where you will find your tutor feedback.
- 4. Download** a copy of your assignment in the current view (with tutor feedback highlighted) or get a copy of your digital receipt (from when you submitted the work).

Viewing feedback

Enabling the Grading layer and disabling the Similarity layer will allow you to see your tutor's feedback on your submission.



Carbon Monoxide: To Boldly Go Where NO Has Gone Before

HLLC5404 SDMeer

1 Citation Needed

Carbon monoxide (CO) is a colorless, odorless, and tasteless gas that is slightly less dense than air. It is toxic to animals that use hemoglobin as an oxygen carrier (both invertebrate and vertebrate) when inhaled in concentrations above about 35 ppm, although it is also produced in normal animal metabolism in low quantities, and is thought to have some normal biological functions. In the atmosphere, it is spatially variable and short lived, having a role in the formation of ground-level ozone.

Carbon monoxide consists of one carbon atom and one oxygen atom, connected by a triple bond that consists of two covalent bonds as well as one dative covalent bond. It is the simplest oxocarbon and is isoelectronic with other triply-bonded diatomic molecules having ten valence electrons, including the cyanide anion, the nitrosonium cation and molecular nitrogen. In coordination complexes the carbon monoxide ligand is called carbonyl. **Good point**

Carbon monoxide is a silent killer. It has no smell, no taste, and no sound. Neither people nor animals can tell when they are breathing it, but it can be fatal. **Weak Transition**

Carbon monoxide (CO) is a by-product of combustion. Common household items, such as gas fires, oil-burning furnaces, portable generators, charcoal grills, among others, put people at risk of exposure to this poison gas.

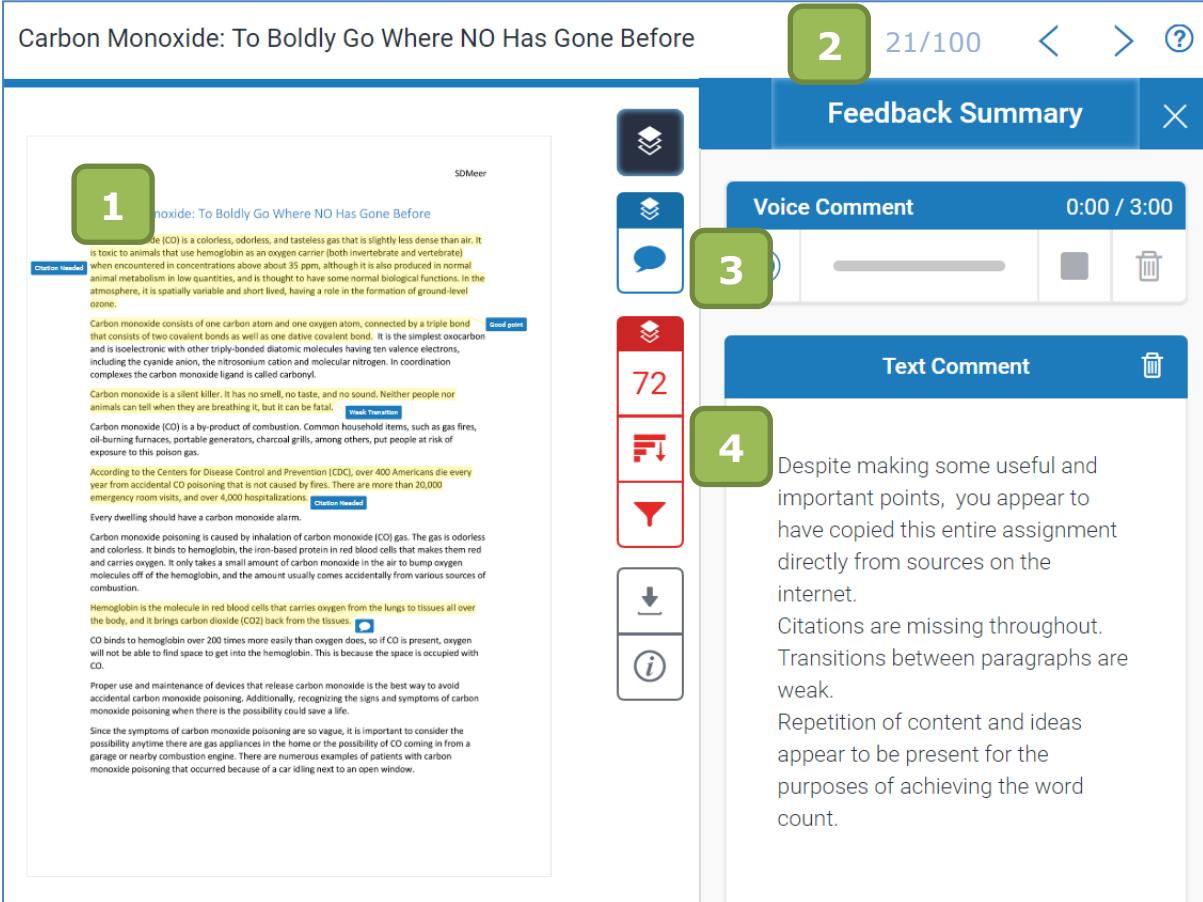
According to the Centers for Disease Control and Prevention (CDC), over 400 Americans die every year from accidental CO poisoning that is not caused by fires. There are more than 20,000 emergency room visits, and over 4,000 hospitalizations. **Citation Needed**

Every dwelling should have a carbon monoxide alarm.

2 **3** **4**

- 1.** Your assignment with tutor comments. Click on each of the blue labels to see further information.
- 2.** Toggle the Grading view on and off.
- 3.** Open the Feedback Summary panel to view additional comments about your complete piece of work. These comments will usually relate to the grading and/or assignment criteria.
- 4. Download** a copy of your assignment in the current view (with tutor feedback highlighted) or get a copy of your digital receipt (from when you submitted the work).

Feedback panel



Carbon Monoxide: To Boldly Go Where NO Has Gone Before

2 21/100 < > ?

Feedback Summary ✕

Voice Comment 0:00 / 3:00

Text Comment ✕

Despite making some useful and important points, you appear to have copied this entire assignment directly from sources on the internet. Citations are missing throughout. Transitions between paragraphs are weak. Repetition of content and ideas appear to be present for the purposes of achieving the word count.

1. Your assignment with tutor comments.
2. Your grade, when it has been released.
3. Some tutors may choose to leave audio feedback on your assignment.
4. Text feedback.

With the tutor comments visible, and the Feedback Summary panel open, you can choose to download a PDF copy of your assignment with all of this information include.

Important things to consider

1. Always read and understand your tutor feedback. You are expected to use this feedback to improve future assignments.
2. Your tutor can help you to understand anything that does not immediately make sense.
3. Visit Skills@Cumbria for additional information and support with:
 - a. Finding information
 - b. Reading and notetaking
 - c. Academic writing
 - d. Referencing
 - e. And much more...